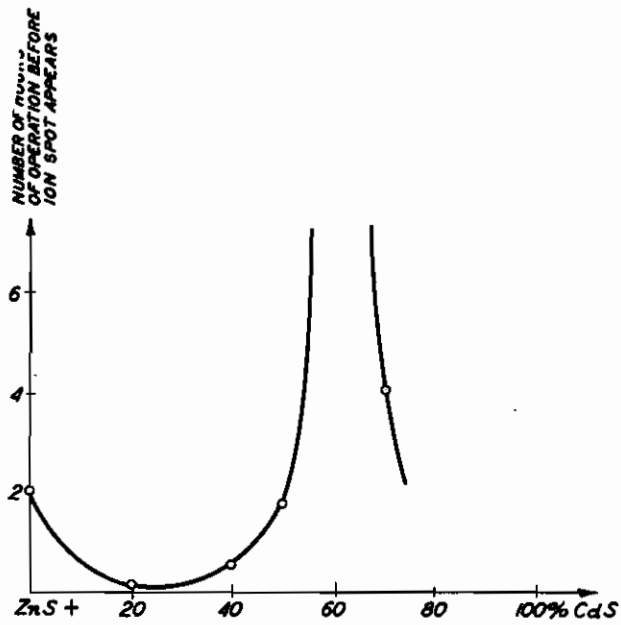


PUBLISHED
MAY 25, 1943.
BY A. P. C.

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LUMINESCENT MATERIAL
Filed Dec. 9, 1941

Serial No.
422,216



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ALIEN PROPERTY CUSTODIAN

LUMINESCENT MATERIAL

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Application filed December 9, 1941

The invention relates to a process for synthesizing luminescent materials and, in particular, is directed to producing luminescent materials of improved properties and with the further advantage that the resistance of the resultant material against loss of luminosity by impacting ions is greatly reduced.

Luminescent materials particularly those adapted to become excited under the influence of electric bombardment such as cathode ray beams in tubes used for television, oscillographs and allied uses have been known for some time by the commonly used luminescent materials known to the workers in the art. These luminescent materials had the drawback that during the operation of the tube a dark spot of larger or smaller dimensions appeared on the fluorescent screen and that this spot became darker and darker with increasing time of operation. The existence of this spot is explained by the fact that an ion stream is produced within the tube and that this ion stream remains undeflected by the magnetic deflecting means so that the ions impact always the same spot of the luminescent screen during the entire period of operation. This ion bombardment reduces the luminescence of the luminescent material.

It is known that zinc-silicate and borium-nitrate are sufficiently insensitive to impacting ions. These materials however can not be used in many cases on account of other undesirable properties. The opinion was up to now that zinc-sulfide and cadmium-sulfide are very sensitive to ion bombardment so that their luminosity is reduced within very short periods of operation.

Accordingly it is one of the purposes of our invention to provide a luminescent material of the zinc-cadmium sulfide type whose sensitivity against the influences of ion bombardment is greatly reduced.

A further object of our invention is to provide a luminescent material of the sulfide type which will emit white light of high intensity under cathode ray bombardment.

Other objects and advantages of our invention will be immediately apparent to those skilled in the art upon reading the following description of our invention.

According to the invention the luminescent material is a crystalline structure comprising the components of the sulfide type in a single crystal lattice structure in such a composition that the components are in stoichiometric equilibrium in the crystal. The luminescent material contains f. i. zinc-sulfide ZnS and cadmium sulfide CdS in

stoichiometric equilibrium. The molecular weight of ZnS is 97.44 and the molecular weight of CdS is 144.47. A zinc cadmium sulfide containing 59.6 weight per cent of CdS is a crystal including one mol CdS for every mol ZnS . A crystalline luminescent material of this nature is particularly safe against ion bombardment.

The figure shows a curve representing by way of example the sensitivity of the luminescent material against ion bombardment in dependency of the CdS content. The ordinate represents the period of time of ion bombardment in hours after which the ion spot appears upon the luminescent screen and the abscissa shows the content of CdS in per cent. The curve has been found by experiments. A luminescent material of 60% CdS does not show any blackening or darkening or loss of luminosity after 20 hours and more of operation.

A luminescent material of this type is by way of example produced in a process shortly described as follows: Zinc-sulfide is precipitated from a solution of zinc-nitrate by sulphuretted hydrogen. In a similar manner cadmium-sulfide is precipitated from a solution of cadmium-nitrate. The two substances are dried and finely pulverised and mixed in the correct proportion so that the two substances are in stoichiometric equilibrium after their combination. The mixture is then placed into a crucible and sulphur is added in sufficient quantity. The charge is heated by any appropriate manner until the substances combine. The charge is then taken out after cooling and the core of the charge is used as luminescent material. A substance for making the crystalline powder not baking may be added.

A luminescent material of the described composition emanates a yellow light. In order to obtain a white colour it is preferable to add a luminescent material of complementary colour. This added component must also be insensitive to ion bombardment. It has been found that a material emitting white light is obtained if 70% of a $ZnCdS$ material with a CdS content of 60% is mixed with 30% pure ZnS . A material of this type is sufficiently insensitive against ion bombardment provided that the mixture is not heated higher than $300^{\circ}C$ after mixing. If the material is heated to a higher temperature a reaction takes place between the components and an ion spot will show after half an hour of operation. This result has a good coincidence with the measurements represented in the figure because by the

reaction a crystal with approximately 40% CdS is produced.

It is however also possible to obtain a material emitting white light which is absolutely safe against ion bombardment by mixing the components in such a manner that the final mixture contains one mol ZnS for every mol CdS. If in such a case a reaction takes place on account of high temperatures a stable crystal will be produced at least upon the surface of the screen so that the ion spot can not appear. Good results have been obtained with a mixture containing 20% ZnCdS with 20% CdS and 80% ZnCdS with 70% CdS. A material of this type shows white light with a very faint strawberry coloured shade. This material has been safe against ion bombardment after 20 hours of operation and more.

It is also possible to change the colour of the emitted light by adding activators so that any desired shades can be obtained.

The invention is not limited to the zinc-cadmium sulfide materials above mentioned which are particularly insensitive to ion bombardment and therefore very suitable as substances for producing luminescent screens of television receiving tubes. The invention relates also to all crystalline luminescent materials composed in a similar manner. In such materials the amount of the components is so chosen that the components are in stoichiometric equilibrium so that the resultant crystal is safe against ion bombardment.

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