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BLEACHING PROCESS FOR FLUIDS

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This invention relates to processes for bleaching or reducing the discoloration of liquids and juices or the like with the aid of hydrogen peroxide.

Basically considered, these processes involve bleaching or decoloring liquids by treating them with hydrogen peroxide vapor. Heretofore prior art bleaching processes wherein hydrogen peroxide was used were operated at reaction conditions at which the hydrogen peroxide was in liquid form, that is the bleaching or decoloring was carried out by simply adding a quantity of hydrogen peroxide to the liquid to be bleached. Although the prior art processes seem satisfactory, simple and fool-proof, they have proven to be unsatisfactory in some cases, particularly in the decoloring of sugar juices before they are concentrated, and in the bleaching of table oils and fats.

The particular difficulty in these cases is that the substances are sensitive and that the resulting product must be peroxide free. Since it is impossible to gauge accurately the amount needed, the excess must be removed by resorting to a further processing. Prior workers have also proposed to decolor liquids with the aid of gaseous sulfuric acid, chlorine or ozone as well as to treat solids with gas-forming hydrogen peroxide.

In accordance with a preferred embodiment of this invention, liquids, such as sugar juices and table oils, are bleached or decolorized by treating them with vaporous hydrogen peroxide. The reaction conditions are such that the process may be carried out in the form of a continuous reaction. It is merely essential that the treatment be carried out in an apparatus which will permit the liquid to be treated to be in the apparatus in an agitated state so as to cause a proportionately large surface area of the liquid to be exposed to vapors containing hydrogen peroxide which are simultaneously present in the apparatus.

In carrying out the process of my invention, it is preferably advisable to make use of apparatus which will permit the use of a continuous process. For example, it is preferable to make use of a spray tower or other tower into which the liquid to be bleached can be introduced at the top and made to pass downwardly therethrough in thin agitated layers which are in contact with vaporous peroxide flowing in countercurrent. A continuous process of this latter type is susceptible of ready control of the degree of bleaching. This control may be exercised by varying the construction of the apparatus itself, that is, by changing the length of the route taken by the material to be bleached. This may be done in

any number of ways, for example, by varying the length of the tower, varying the amount and type of fillers (Raschig rings, plates, etc.) or by the variation of corresponding built in irrigation surfaces.

The control of the bleaching process may also be affected by other means, for example, by the presence of various catalysts in the tower, for example: rings, plates etc. with surfaces made of metals or chemicals which are catalysts for H_2O_2 (Fe, Ni, Co, Mn, Ag, Pt, MnO_2 , Co_2O_3 , Fe_2O_3 , etc.) and/or adding such chemicals ($CO(NO_2)_2$, $CoSO_4$, $CuSO_4$, $NiSO_4$, $Fe(NO_2)_2$ etc.) in small parts to the liquid to be bleached, by the respective concentrations of the liquids to be bleached and of the bleaching agent present, by varying the speeds at which the material to be bleached passes through the tower and by regulating the temperatures of the tower, of the material to be bleached and of the bleaching agent. The reaction may also be influenced by pressure conditions in the apparatus. However, although either superatmospheric or subatmospheric pressures are preferably used, the process may also be carried out at normal pressures.

The temperature conditions may be varied over a wide range. For example, in another preferred embodiment of my invention, the temperatures of the various mediums may be regulated in such manner that a fractional condensation of the bleaching agent on thin layers of bleaching material takes place in the upper portion of the tower, while the temperatures of the lower portion of the tower and the mediums are so regulated that any excess in bleaching agent present in the material to be decolorized is again vaporized therefrom in the lower portion of the tower. It is thereby possible to have the bleaching agent react in liquid form on the material to be bleached during the first step of the process regardless of the concentration of the vapors while the previously condensed excess of bleaching agent which reaches the lower portion of the tower vaporizes therefrom and increases the concentration of the bleaching vapor entering the apparatus. This process guarantees a dependable and rational execution of the decoloring process even in the most difficult cases.

It has been found that by using this process even very sensitive liquids, such as table oils, liquid fats, and the like, may be bleached without harm because only enough peroxide is withdrawn from the bleaching vapors as is required to decolor or bleach the material.

It is therefore possible, by adjusting the reaction conditions in the apparatus, to take care of maximum discoloration, to use this process to decolor satisfactorily sensitive liquids whose discoloration may vary over a wide range. For example, it is possible by using this process to decolor satisfactorily thickened sugar juices which leave the boilers with all shades of discoloration.

The bleaching agent, which is introduced into the bottom of the apparatus and passes up in countercurrent and reaches the top of the tower, may be withdrawn therefrom by suction or some other suitable means, accumulated and used again after its concentration has been brought

up to that the bleaching vapor initially introduced into the tower. In this manner, it is possible to carry out the process in a rapid, satisfactory, economic matter without loss of bleaching agent, and without harm to the materials bleached.

It will be understood that the embodiments of my invention described in this specification are only specific examples of the processes utilizing the principles of the invention. It is therefore desirable that the invention be interpreted broadly within the scope and spirit thereof.

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